

SPLIT SAMPLING EVENT DATA REVIEW AND DISCUSSION

Data collected at the Homestake Mining
Company Grants site by USGS and
HMC/Arcadis in 2016

April 30, 2018

Meeting Agenda

- 10:00 – 10:10 Building Safety Overview, Introductions, and Safety Share
- 10:10 – 10:20 Meeting Objectives
- 10:20 – 11:30 Arcadis Presentation 2016 Split Sampling Results, Passive
Sampler Data, and Geophysics
- 11:30 – 12:30 USGS Presentation 2016 Sampling Results and Geophysics
- 12:30 – 1:00 General Discussion

Arcadis presentation

- H&S moment
- HMC Grants Mill site overview
- Split sampling results and water chemistry
- Passive samplers and bench testing
- Geophysics
- Wells DD-3, DD-4, DD-5
- Conceptual Site Model revision, geological logging, and soil sampling
- Conceptual Site Model for groundwater

Health and Safety Moment

Wash your car regularly

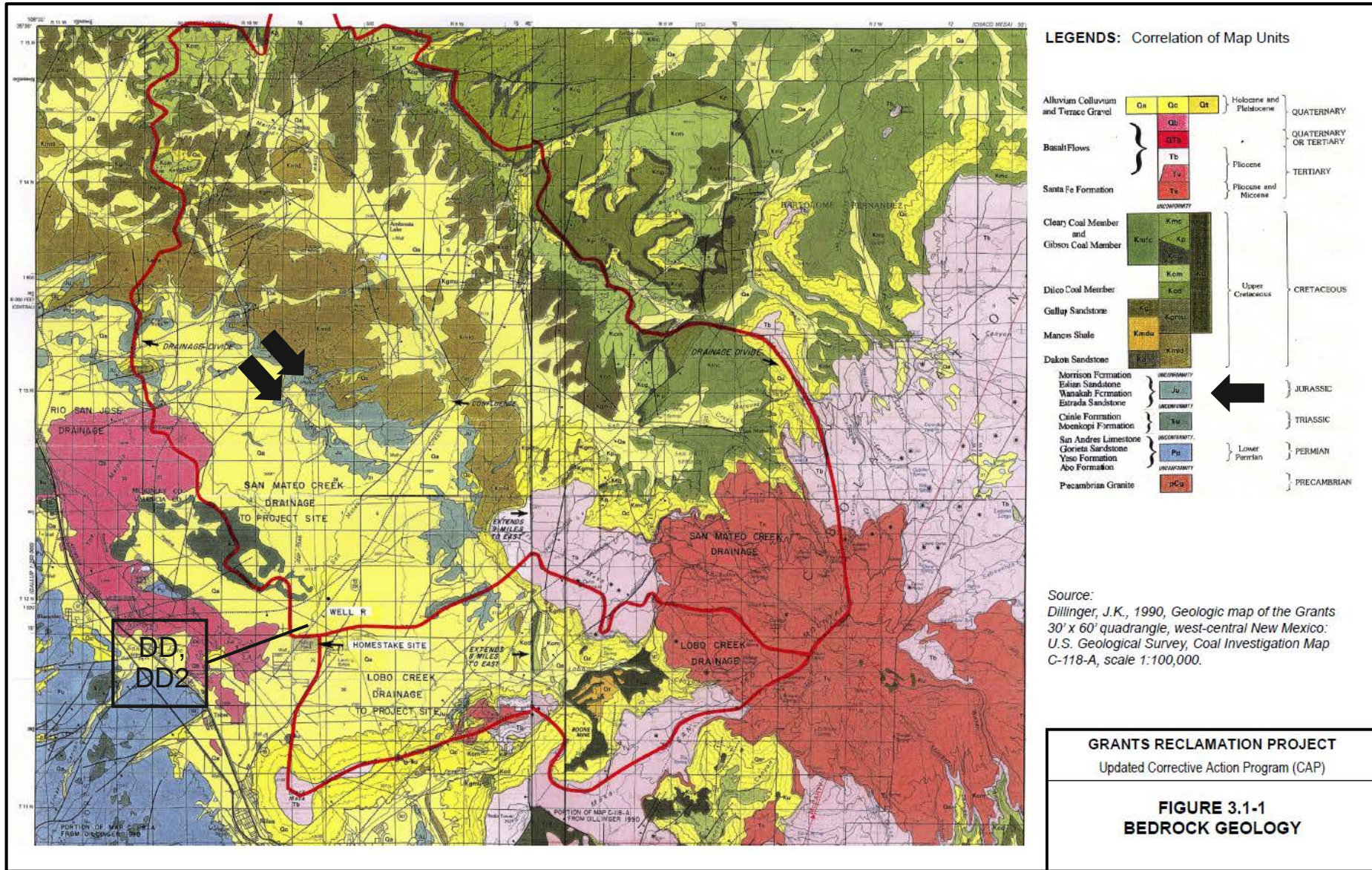
Hazards resulting from a dirty car

- Poor visibility for driver
 - Dirty windows
 - Angle of sun
 - Headlights
- Poor visibility for other drivers
 - Obscured headlights, tail lights, and blinkers
- Mechanical health and longevity
 - Road salts = corrosion

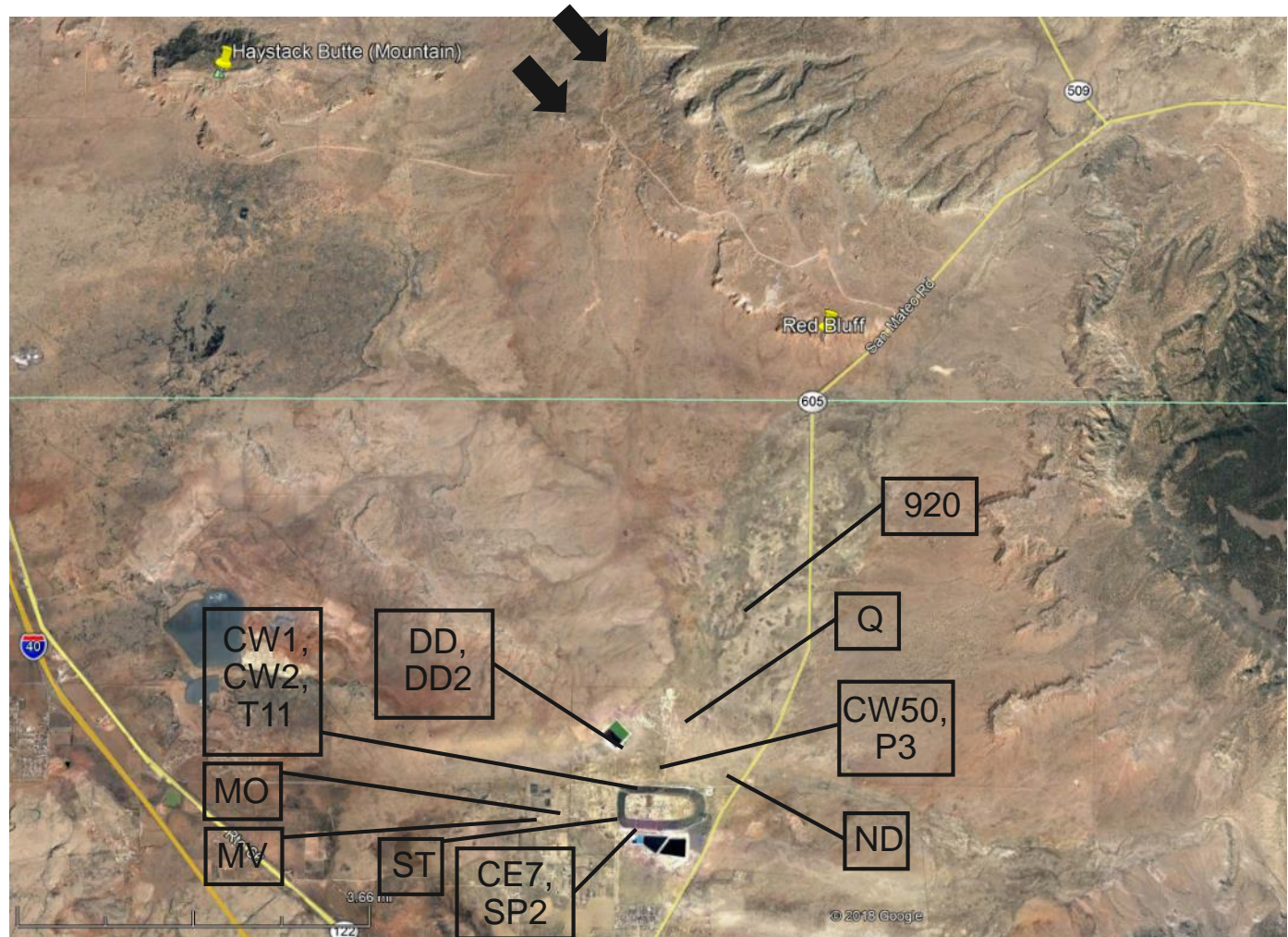


HMC Grants Mill site

San Mateo Creek Basin Geology



San Mateo Creek Basin Geology



San Mateo Creek Basin Geology

Alluvium is from
eroded highlands

This rock contains
ore-grade uranium

Results in
disseminated
uranium-
containing
particles in
alluvium

Erosion/fluvial
deposition is a
heterogeneous
process (visible)



Split sampling event: Summer 2016

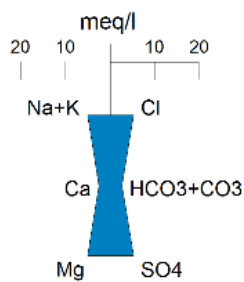
Comparison of Split Sampling Results – Available Data

Samples collected via 3 methods: volumetric, micropurge, passive sampler

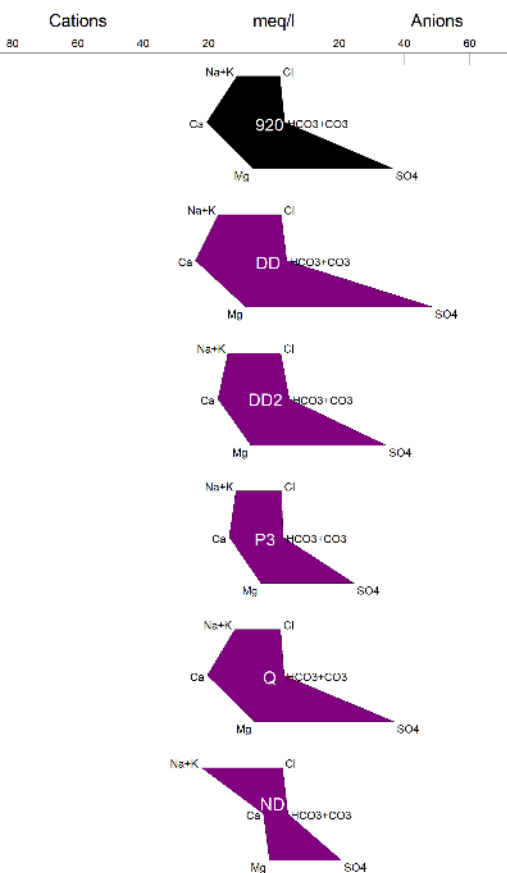
- ☐ Missing all passive sampler data
- ☐ Metals:
 - ☐ missing total uranium from DD, DD-2, MV, ND, Q, T-11 volumetric purge samples
 - ☐ missing dissolved uranium from T-11 volumetric purge sample
 - ☐ missing all uranium from DD, DD-2, MV, ND, Q, T-11 micropurge samples
 - ☐ missing selenium from DD micropurge sample
- ✓ Major anions and cations
- ✓ Nitrogen compounds
- ✓ Alkalinity
- ✓ Total organic carbon
- ✓ Radionuclides
- ✓ Isotopes
- ✓ Dissolved gases (CFCs)
- ✓ Geophysical data
- ✓ Field parameters
- ✓ Field Hach analyses: dissolved oxygen and ferrous iron

Split sampling data and major water chemistries

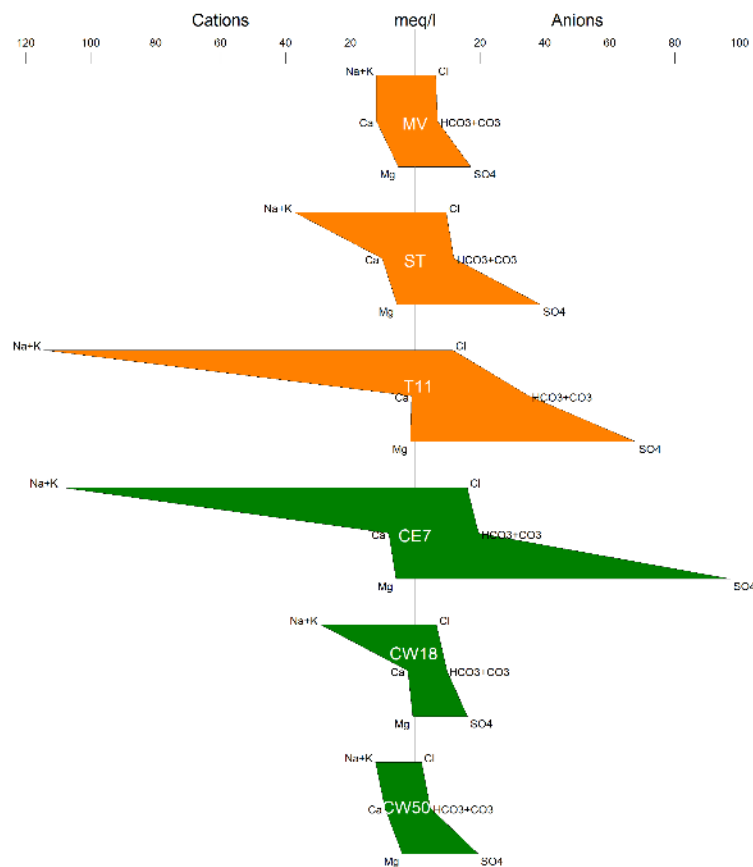
Cations Anions



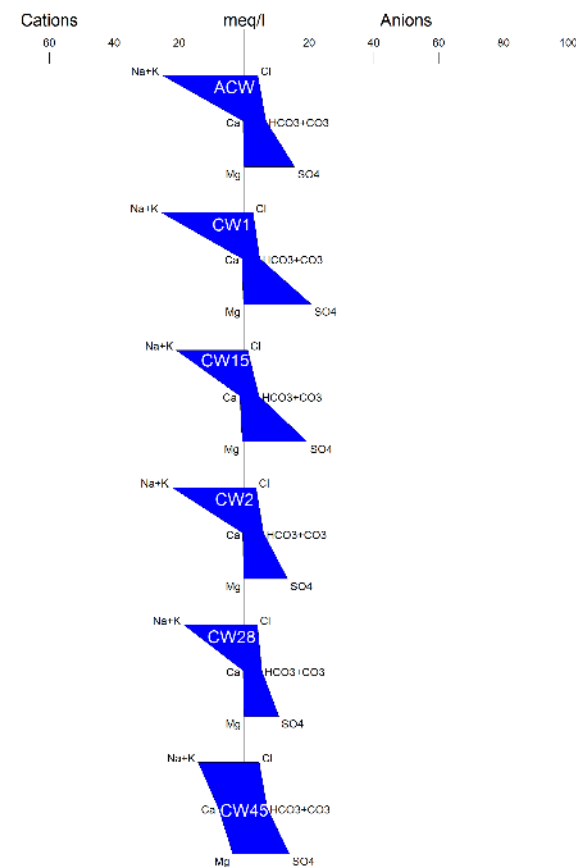
Stiff Diagrams: Upgradient wells

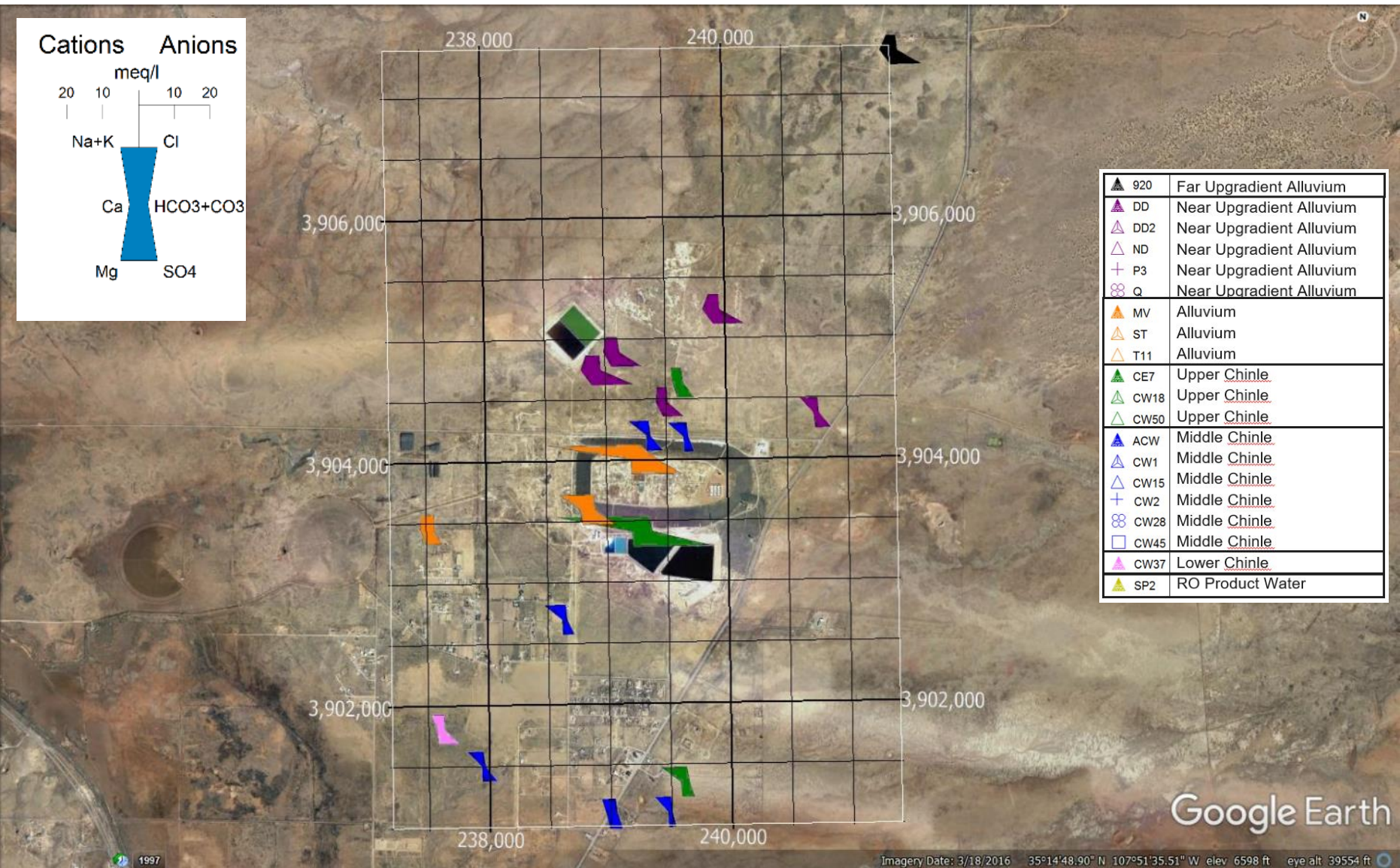


Alluvium (orange) and Upper Chinle (green)



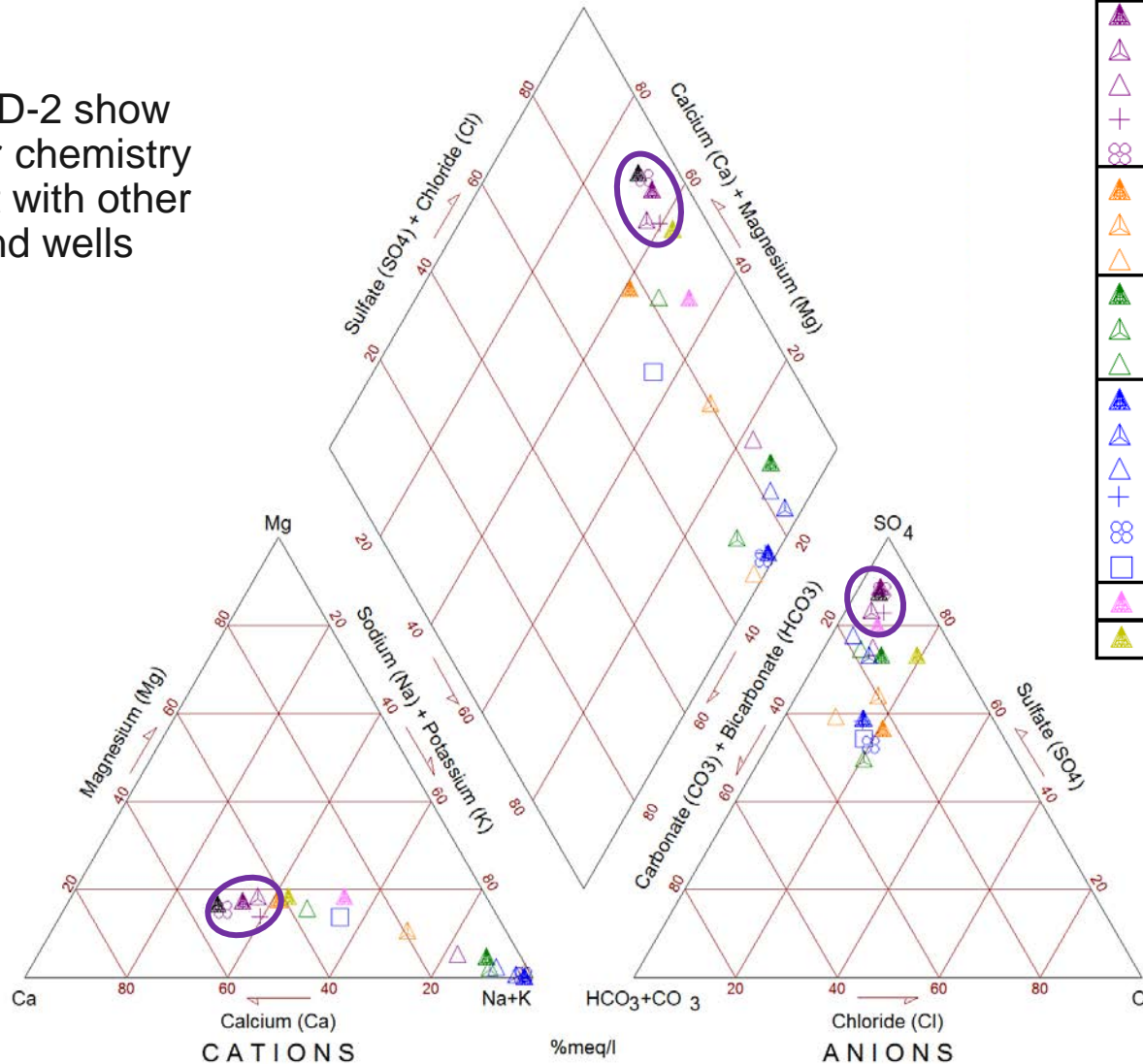
Middle Chinle





DD and DD-2 group
with far upgradient
well 920

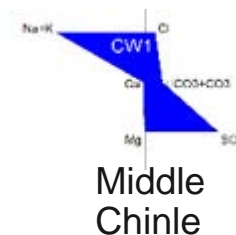
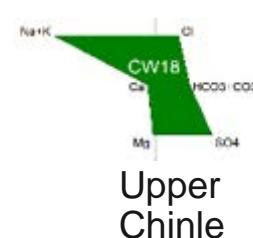
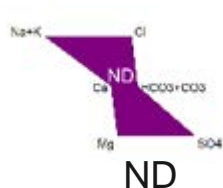
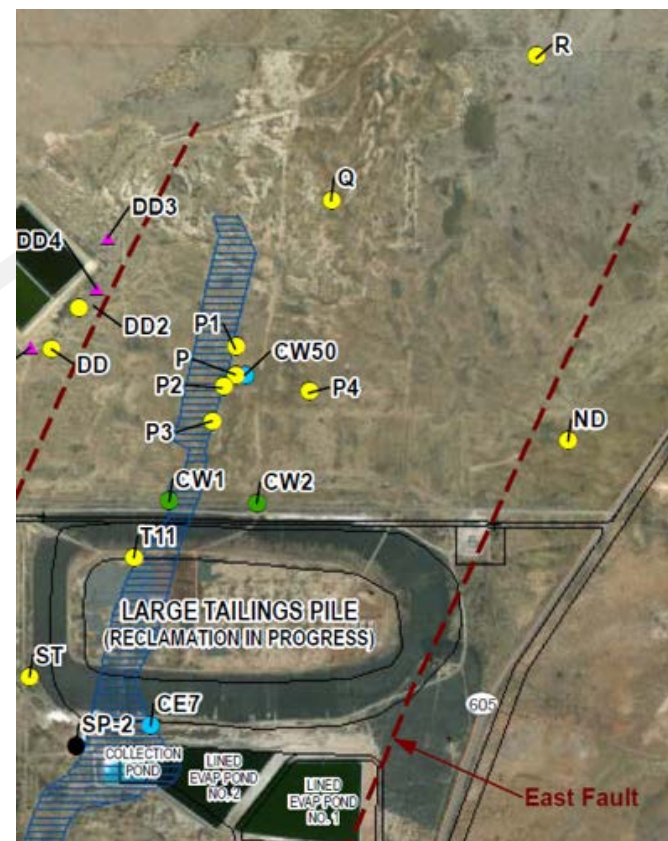
DD and DD-2 show
bulk water chemistry
consistent with other
background wells



▲ 920	Far Upgradient Alluvium
▲ DD	Near Upgradient Alluvium
▲ DD2	Near Upgradient Alluvium
▲ ND	Near Upgradient Alluvium
+ P3	Near Upgradient Alluvium
⊗ Q	Near Upgradient Alluvium
▲ MV	Alluvium
▲ ST	Alluvium
▲ T11	Alluvium
▲ CE7	Upper Chinle
▲ CW18	Upper Chinle
▲ CW50	Upper Chinle
▲ ACW	Middle Chinle
▲ CW1	Middle Chinle
▲ CW15	Middle Chinle
+ CW2	Middle Chinle
⊗ CW28	Middle Chinle
□ CW45	Middle Chinle
▲ CW37	Lower Chinle
▲ SP2	RO Product Water

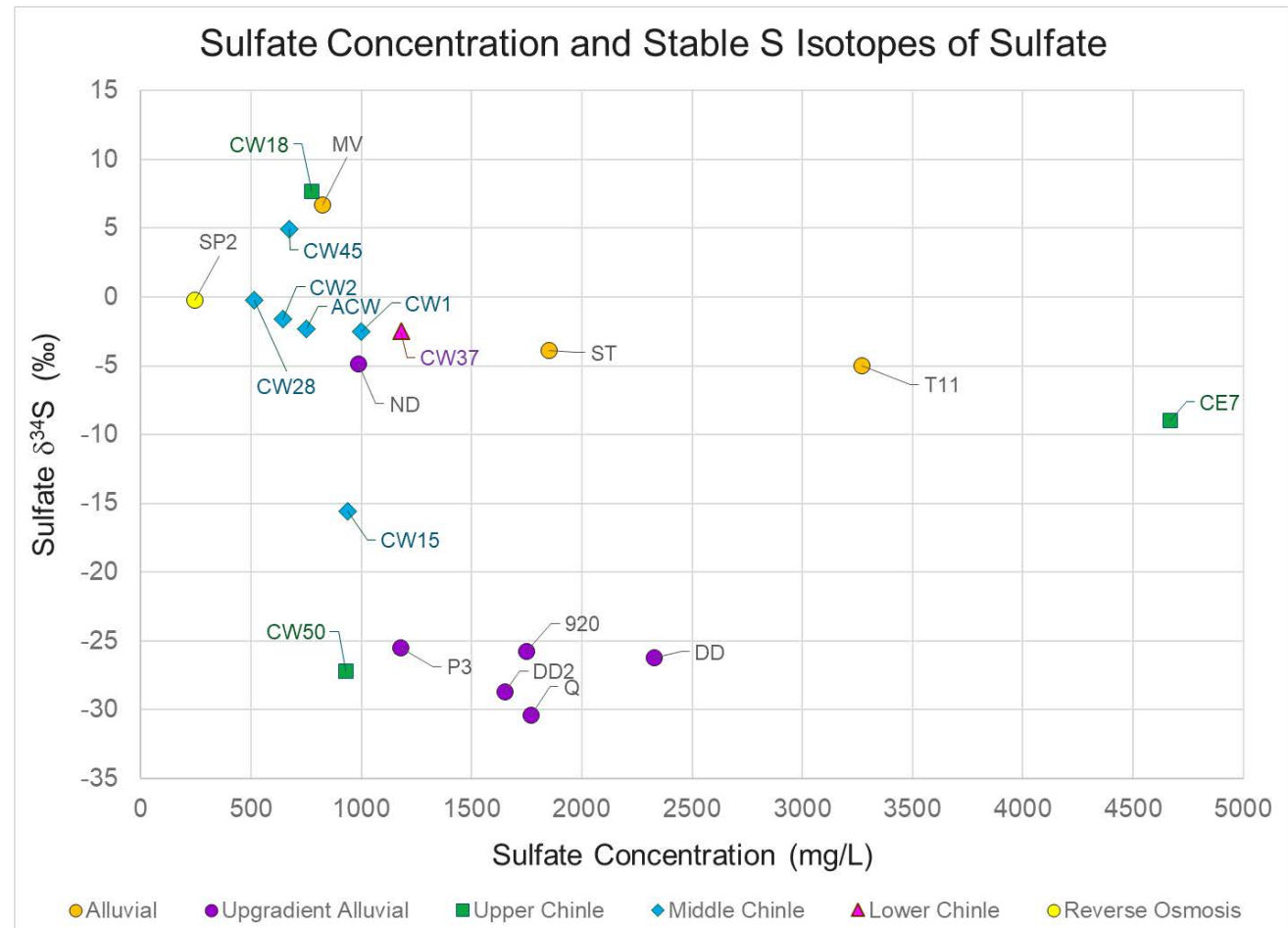
Well ND

- Location: Eastern side of alluvial channel
- Geology: alluvium to 65 ft, underlain by shale
 - Log describes sands and gravelly sands
- Well construction: depth 70 ft, 4-inch, DTW 47.63 ft
- Screen interval: 50-70 ft, across sands, Chinle Shale
- Geochemistry: Has similar piper plot trend as many wells screened in the Chinle Aquifers
- Uranium:
 - Relatively uniform trace for soils with similar peak heights as per DD and DD2
 - Low U concentration in groundwater
- Observations:
 - Well appears to be a geochemical outlier when compared to other background wells
 - Geological/hydrogeological interaction other than just alluvium must be occurring in this area



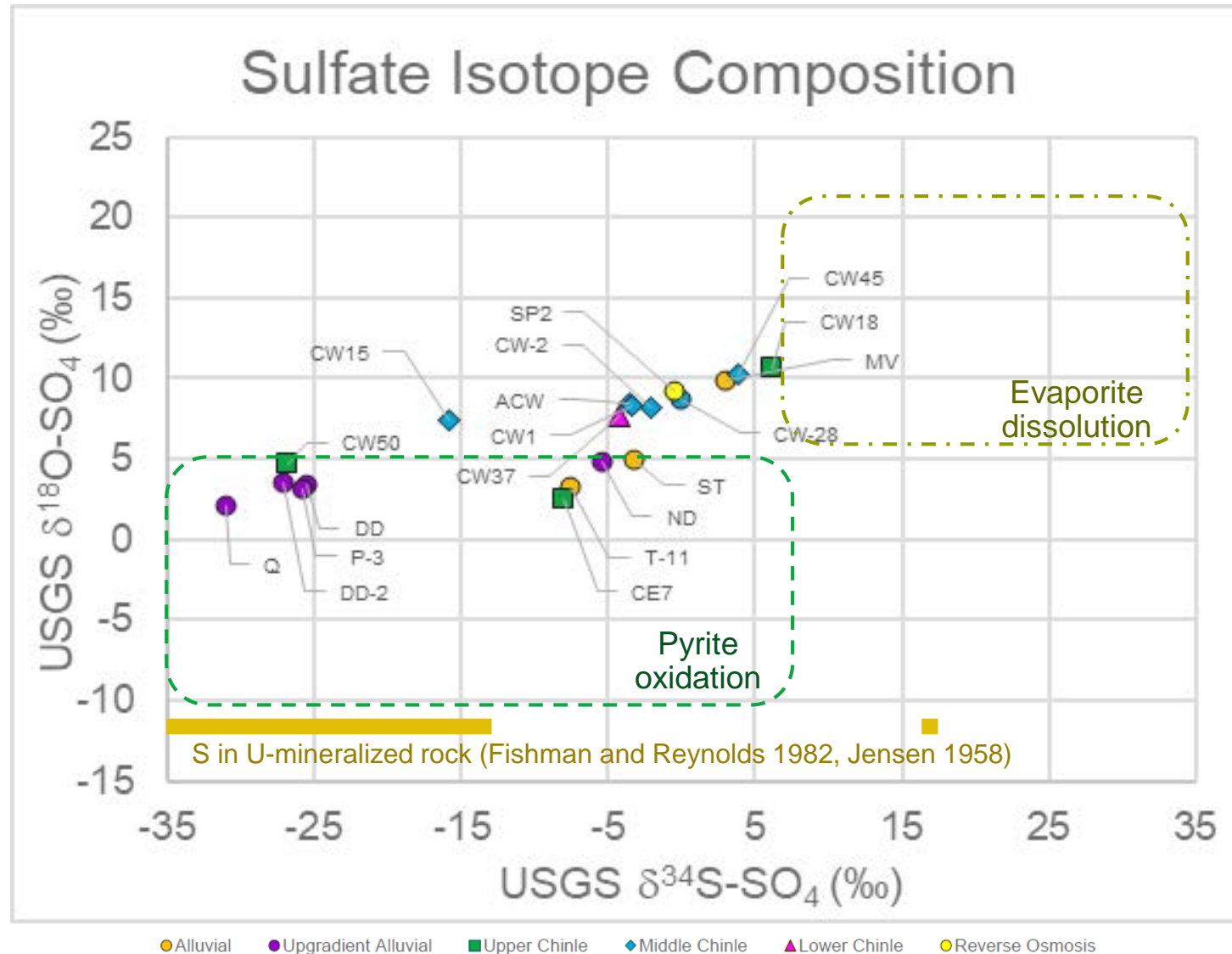
Sulfate S and O Isotopes

- Upgradient alluvial wells and CW50 plot separately from LTP wells T11, CE7, and ST. Stable sulfur isotopes suggest pyrite oxidation is primary sulfate source.
- Samples with lower sulfate concentrations and more positive $\delta^{34}\text{S}$ values suggest sulfate derived from combination of pyrite oxidation and gypsum dissolution.
- Use of sulfuric acid in U ore processing ($\delta^{34}\text{S} \sim -8$ to 32‰) may account for the higher sulfate concentration in LTP wells T11 and CE7



Sulfate S and O Isotopes

- Region contains naturally-occurring sulfide minerals (pyrite) in U-mineralized rocks and U-mineralized sediment present in alluvial aquifer
- Oxidation of S from sulfide minerals results in sulfate formation
- U-mineralized samples from Morrison Fm (sulfide minerals) $\delta^{34}\text{S}$ -29.4 to -41.6‰ (Fishman and Reynolds 1982)
- Sulfide minerals associated with sandstone-type U-deposits $\delta^{34}\text{S}$ -13 to -41‰, one sample +17.6‰ (Jensen 1958)



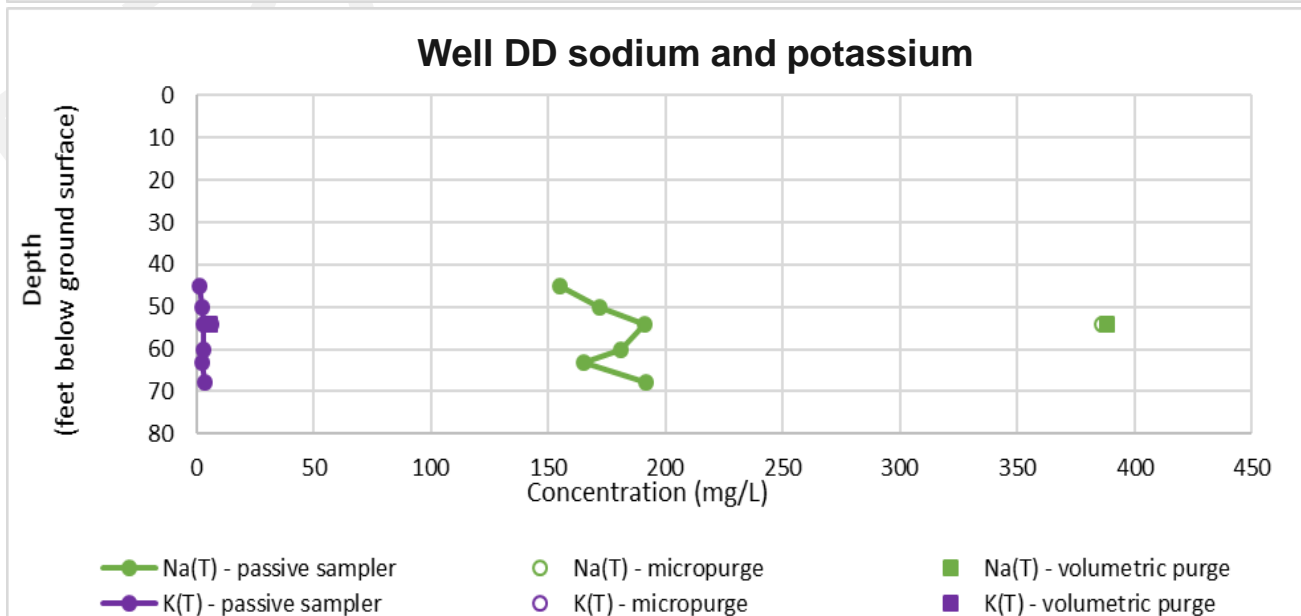
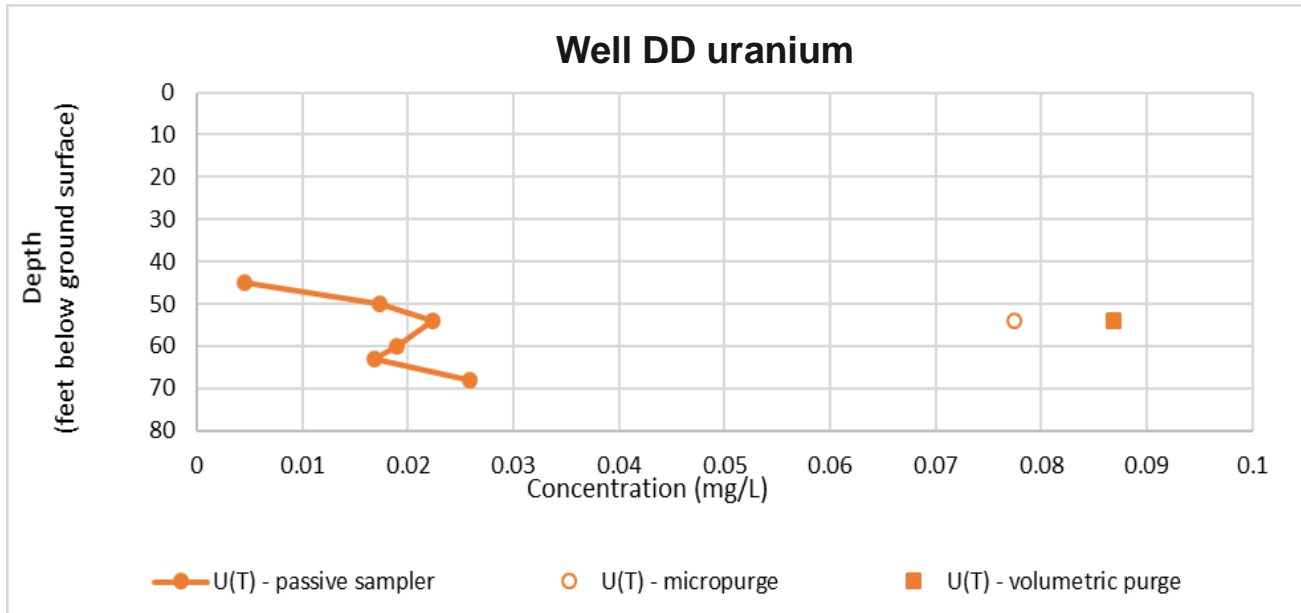
Results by sample method

- volumetric purge
- micropurge
- passive samplers

Passive sampler <<
micropurge or volumetric
purge

Passive sampler ≠
micropurge at same depth

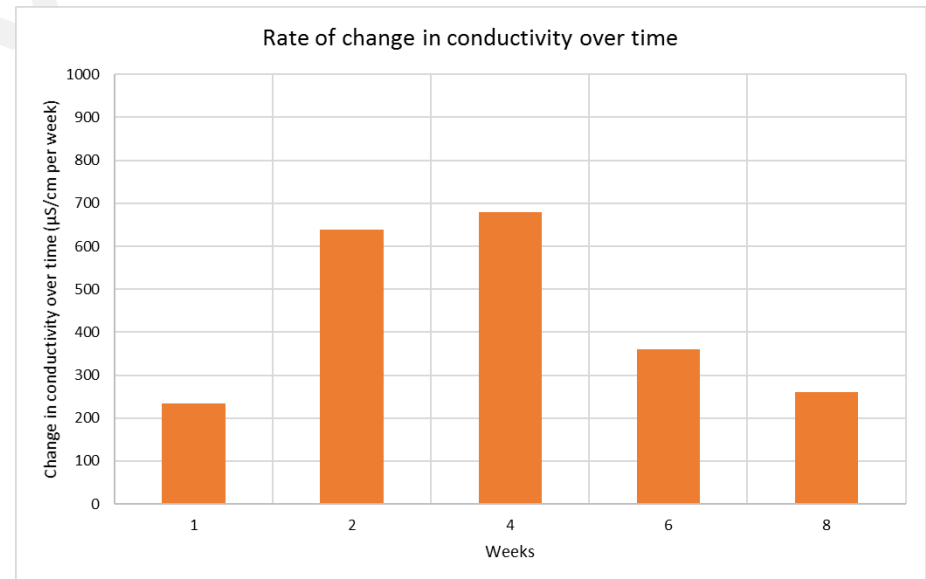
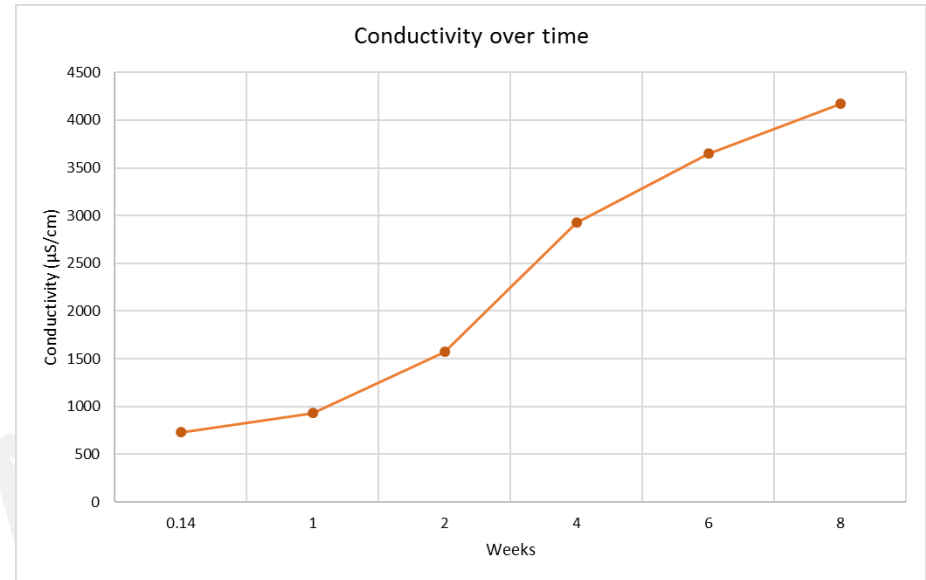
Conservative ions did not
equilibrate



Passive Samplers – bench testing

Passive sampler bench testing results

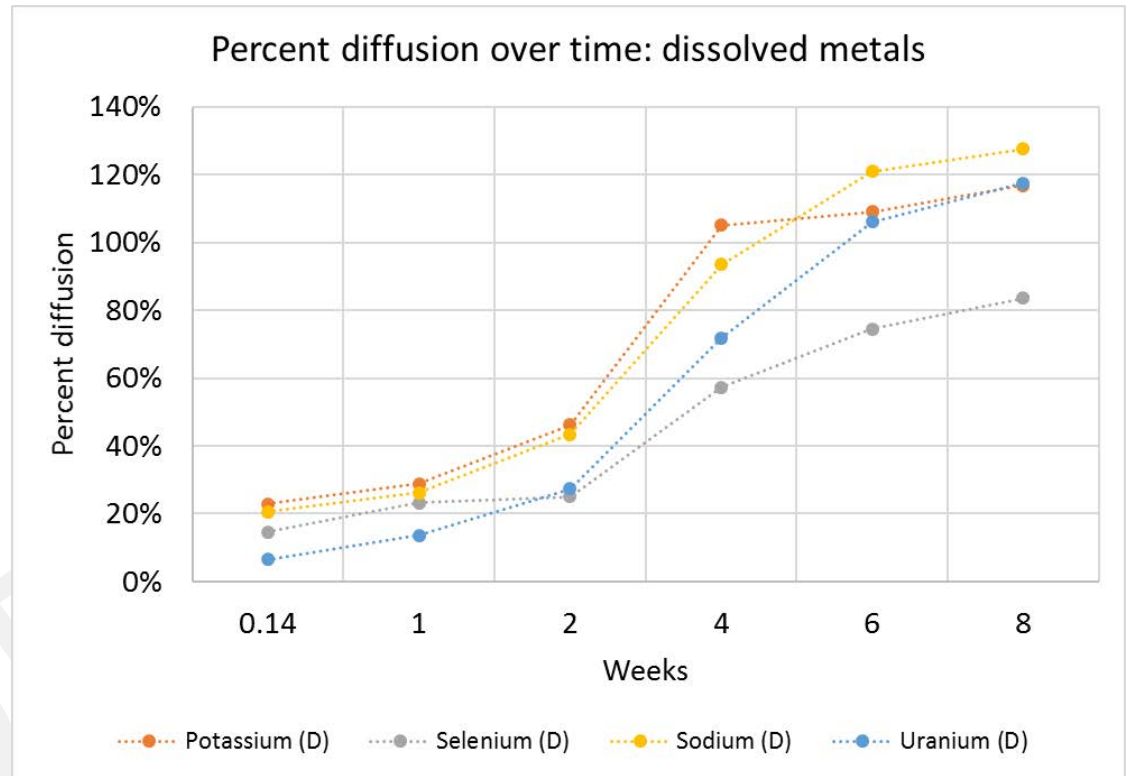
- Conductivity was still increasing in the collected passive samplers at 8 weeks' time
- Peak change in conductivity in the passive samplers occurred at 4 weeks
- Reflects that highest mass flux was occurring around when passive samplers were collected in the field



Passive sampler bench testing results

It takes at least 6 weeks, and likely 8+ weeks, for equilibration to occur

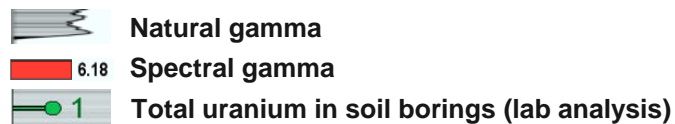
We also saw binding to the nylon mesh, including up to 5 mg/kg uranium



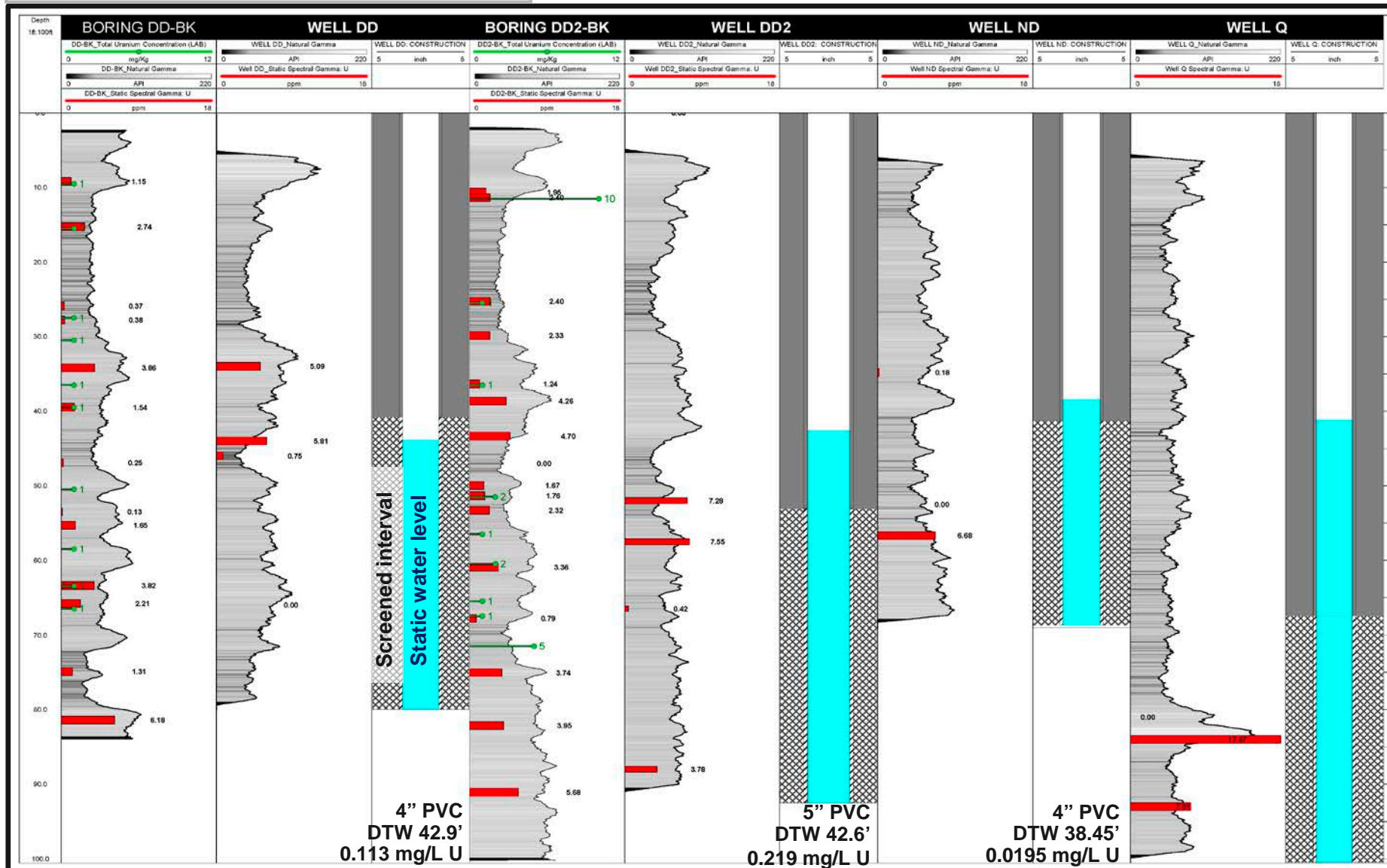
	Fully mixed solution	24hr	Week 1	Week 2	Week 4	Week 6	Week 8
Analyte	% diffusion	% diffusion	% diffusion	% diffusion	% diffusion	% diffusion	% diffusion
Dissolved Metals by Method SW6010B for Na and K, SW6020 for Se and U (mg/L)							
Potassium	100%	23%	29%	46%	105%	109%	117%
Selenium	100%	15%	23%	25%	57%	75%	84%
Sodium	100%	21%	26%	43%	94%	121%	128%
Uranium	100%	6%	14%	27%	72%	106%	117%

Geophysics

Alluvial uranium and well construction

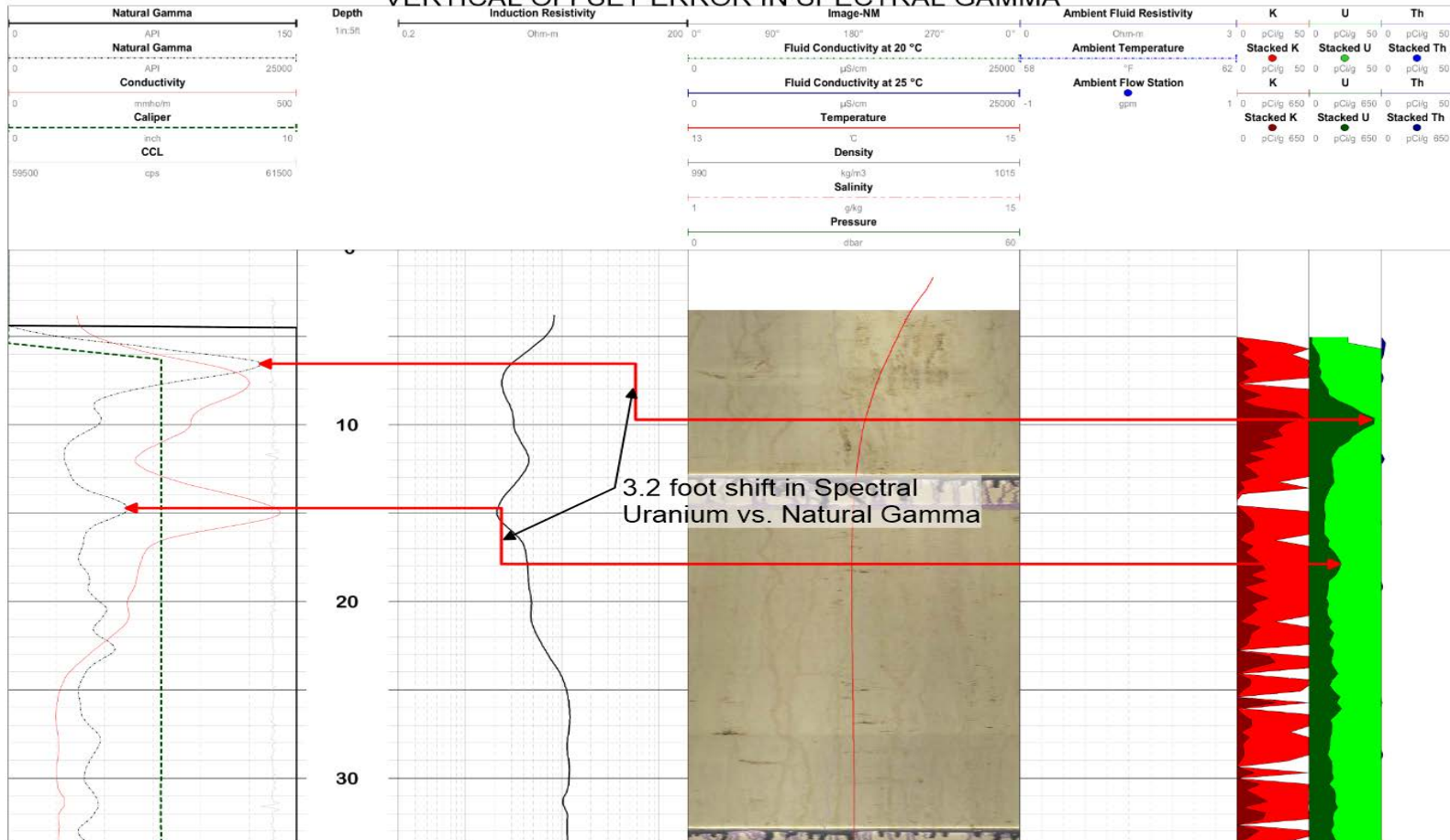


Uranium in alluvium is preferentially in fine grained sediments and varies significantly by location



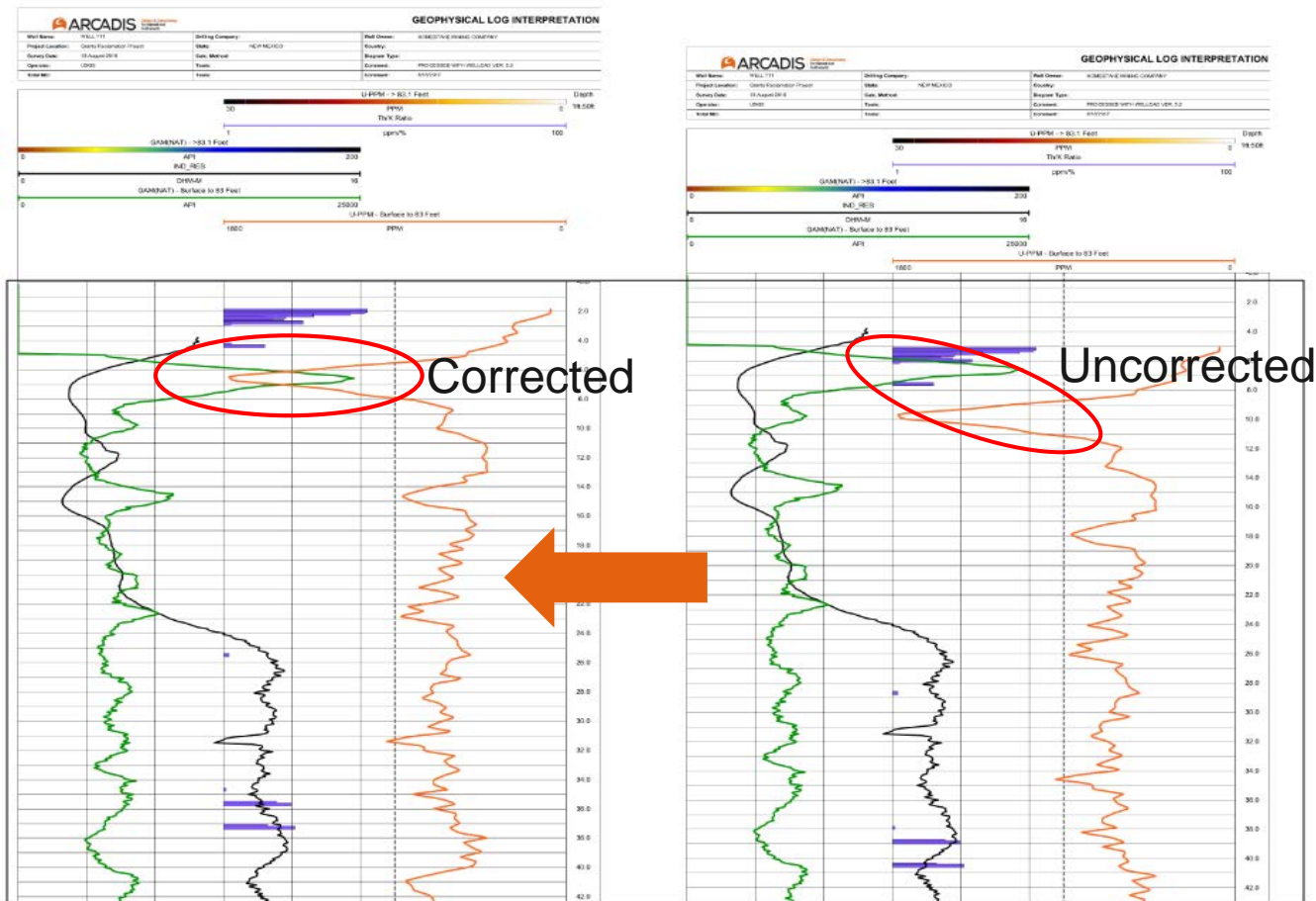
Well T11: Spectral Gamma Shift Error

WELL T-11. USGS, AUG. 2016
VERTICAL OFFSET ERROR IN SPECTRAL GAMMA



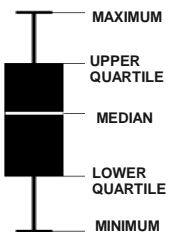
Is spectral gamma correctly aligned at the other wells?

Well T11: Spectral Gamma Shift Error

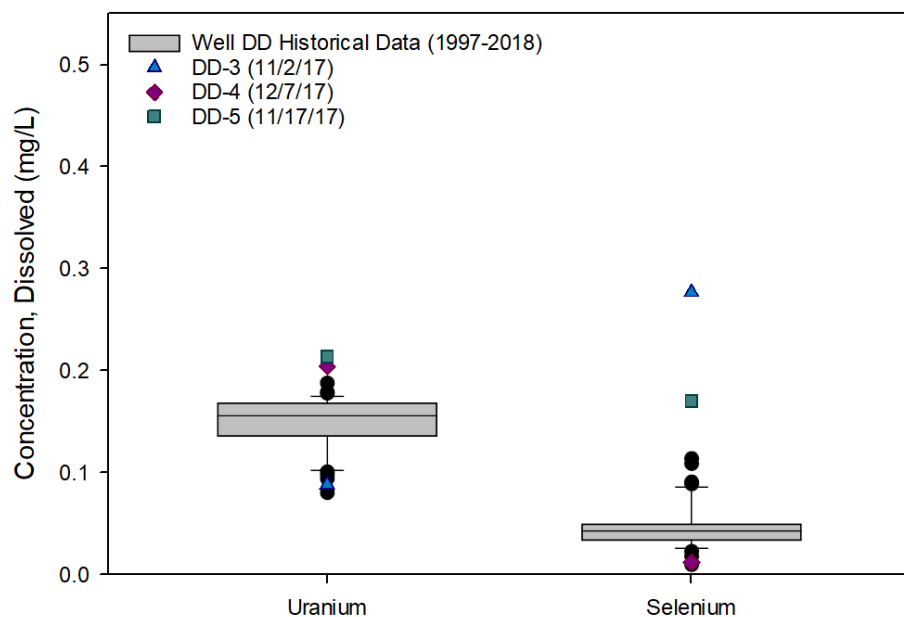


Is spectral gamma correctly aligned at the other wells?

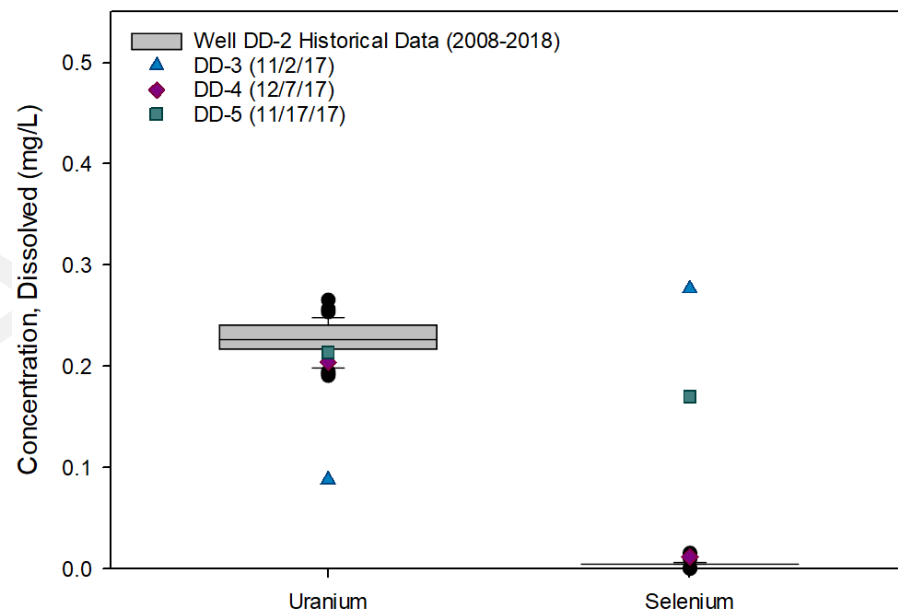
Wells DD-3, DD-4, and DD-5



Well DD



Well DD-2

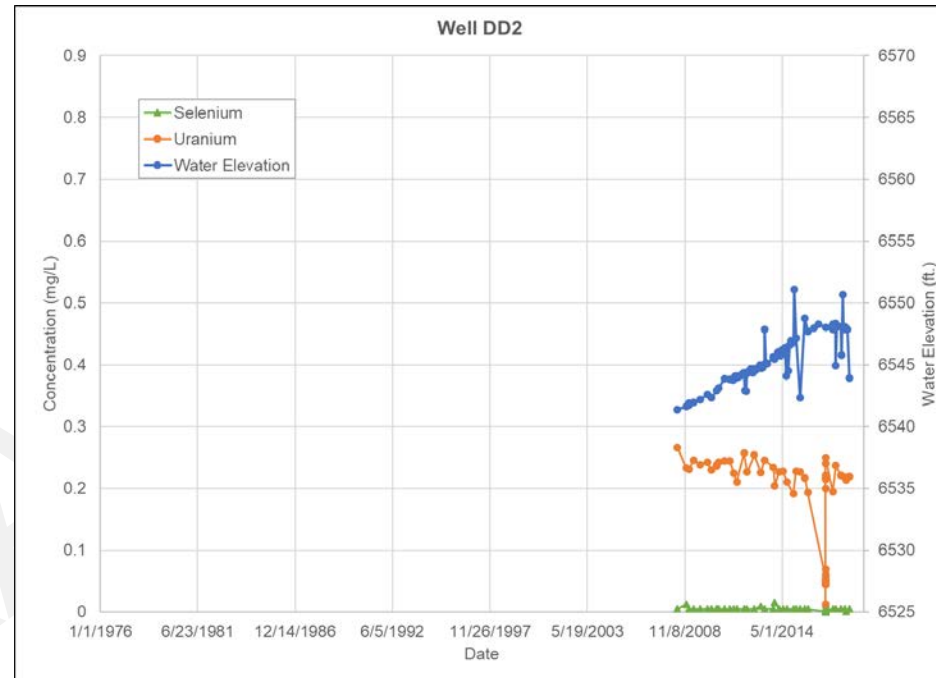
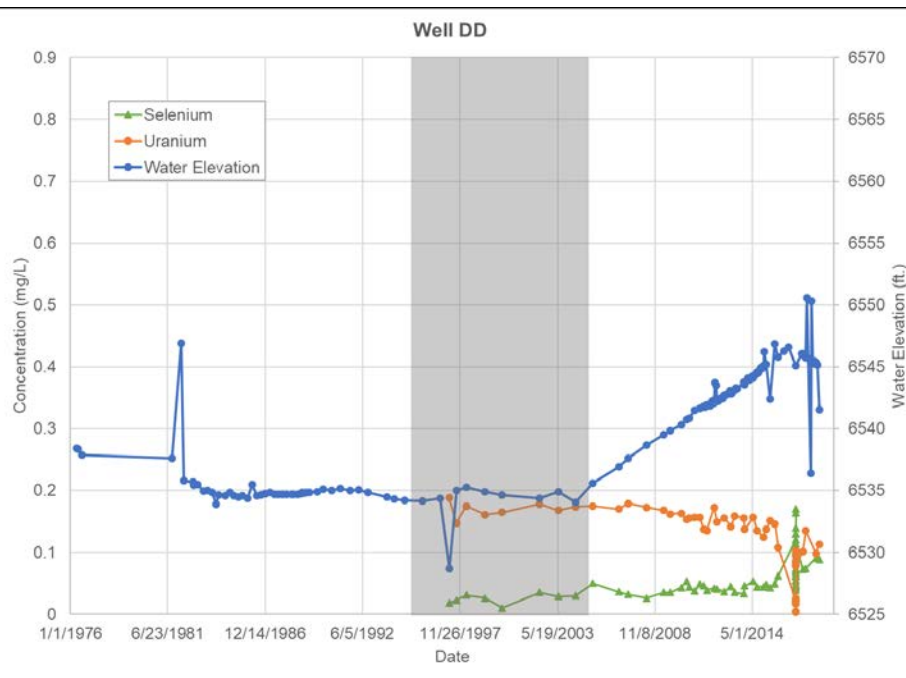


	U	Se
DD-3	0.0884	0.277
DD-4	0.204	0.012
DD-5	0.214	0.170

DD-6 and DD-7 are dry



Changes over time



- Water levels have increased, starting after collection of data for the background study in 2004
 - Due to hydraulic barrier emplaced on north side of Homestake Mill
- Selenium has increased in DD
- Uranium is decreasing in DD and DD2 since the background study

Overview of Conceptual Site Model for Upgradient Concentrations of Uranium in Groundwater:

**Geophysics, and soil chemistry at DD-
BK and DD2-BK**

Location of boreholes



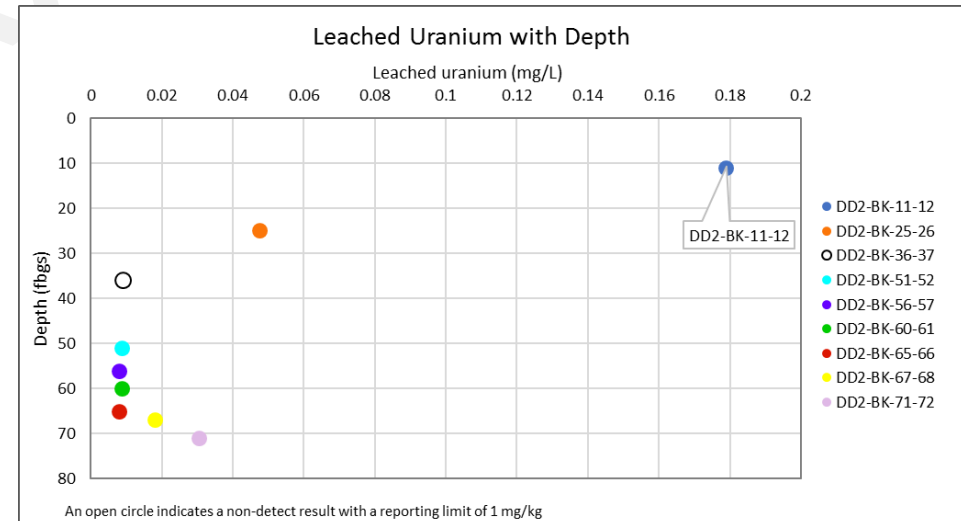
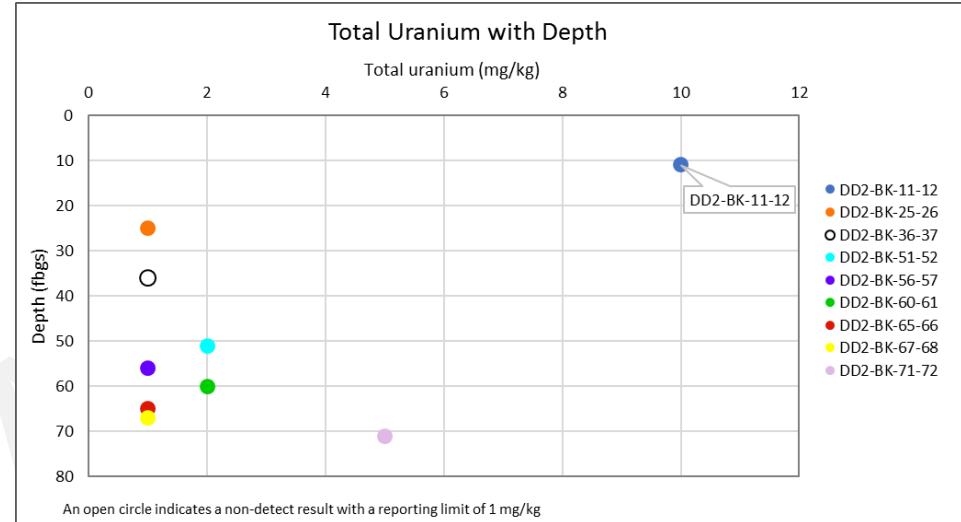
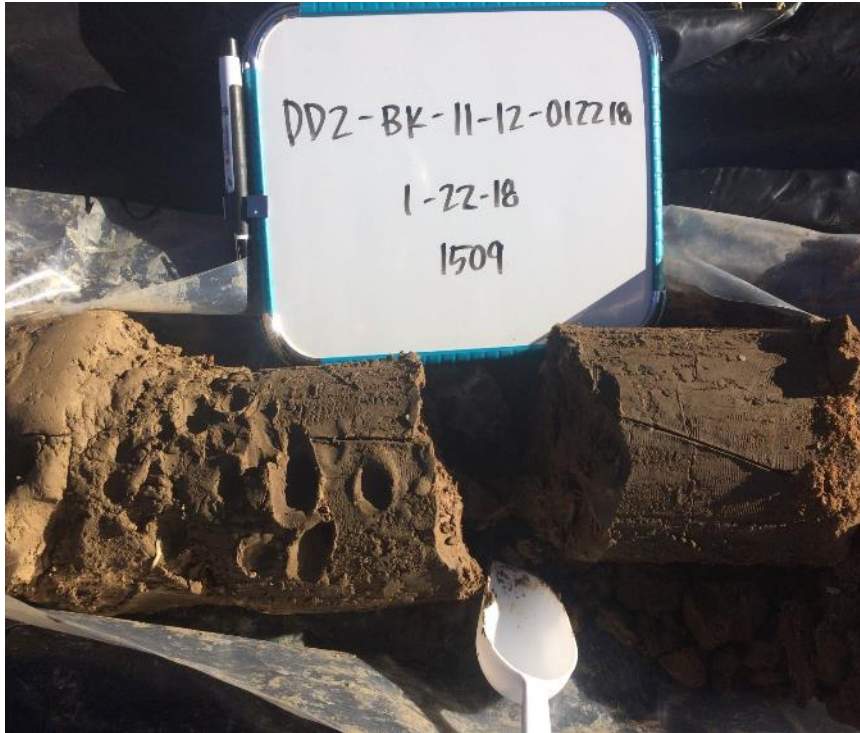
Samples with elevated uranium

Sample ID	Alluvium zone	Total uranium concentration (mg/kg)	Alkaline SPLP leached uranium (mg/L)	Field-logged lithology	ACZ Particle Size Analysis Lithology	DCM analysis
DD2-BK-11-12-012218	unsaturated	10	0.179	CLAY	Clay	Yes
DD2-BK-71-72-012318	saturated	5	0.0305	Gravelly SAND with silt	Sand	Yes
DD2-BK-51-52-012318	saturated	2	0.0086	Silty SAND	—	Yes
DD2-BK-60-61-012618	saturated	2	0.0086	CLAY with trace sand	—	Yes
DD2-BK-25-26-012218	unsaturated	1	0.0477	SAND with trace silt	Sand	Yes
DD2-BK-56-57-012318	saturated	1	0.0079	Silty SAND	—	No
DD2-BK-65-66-012318	saturated	1	0.0080	Sandy SILT	—	No
DD2-BK-67-68-012618	saturated	1	0.0180	CLAY	—	No
DD-BK-36-37-012518	unsaturated	1	0.0127	CLAY	Clay	Yes
DD-BK-58-59-012618	saturated	1	0.0032	CLAY	—	Yes
DD-BK-9-10-012518	unsaturated	1	0.0022	CLAY with trace sand	Clay	Yes

19 samples (excluding duplicate) were analyzed by ELI, only those with detectable total uranium concentrations are shown in the table

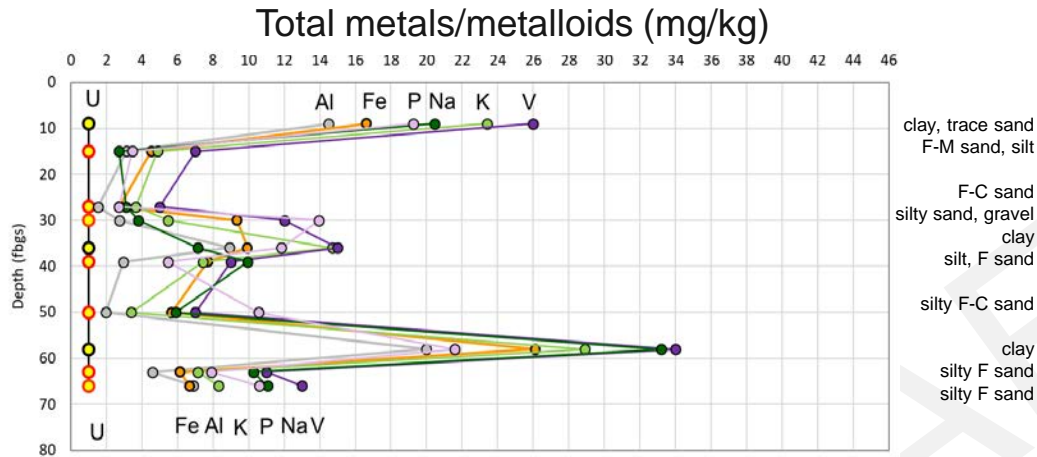
DD2-BK-11-12-012218

- Total uranium concentration: 10 mg/kg
- Alkaline SPLP leached uranium: 0.179 mg/L
- Lithology: Clay

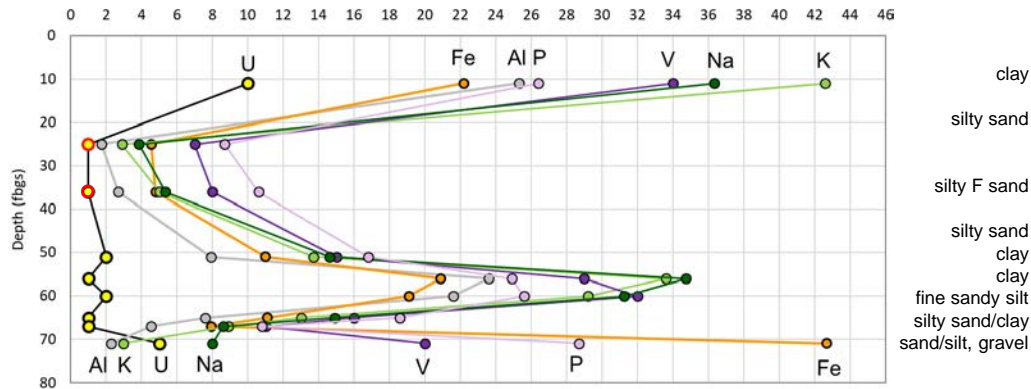


Soil chemistry – total metals

DD-BK



DD2-BK

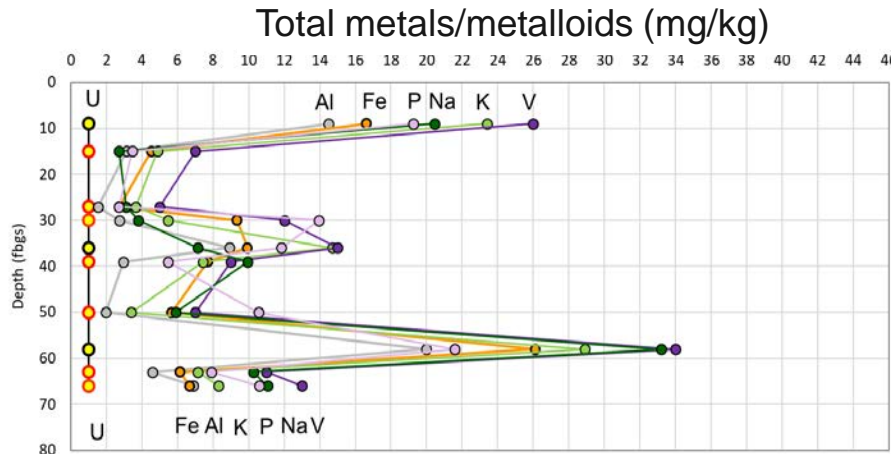


Red circle = non-detect; reporting limit shown
Total Al and Fe/1000; K/100; Na, P/20

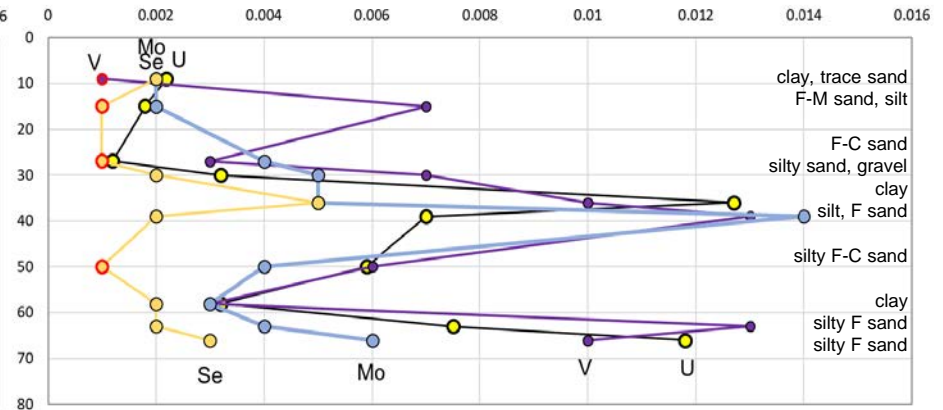
Soil chemistry – total metals and leaching

Leached in an alkaline (simulated groundwater) extraction

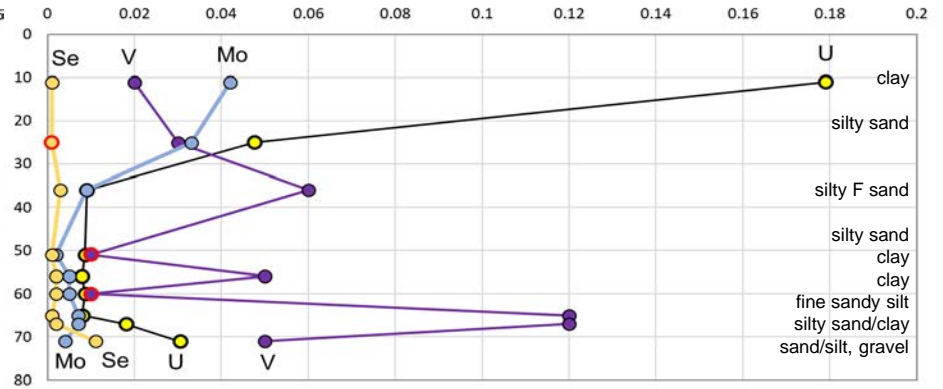
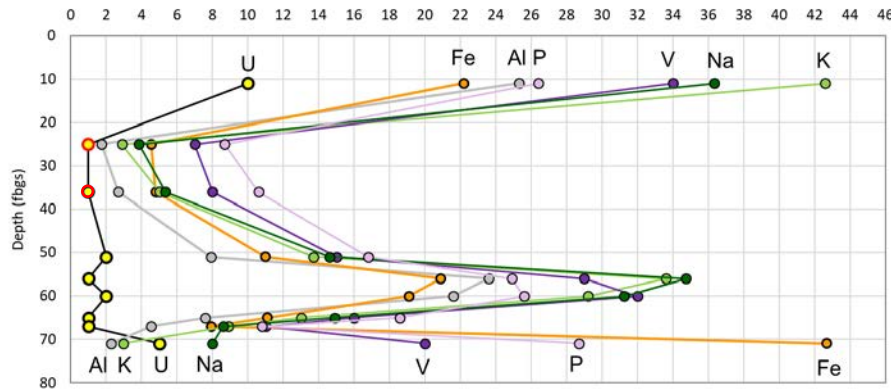
DD-BK



Leached metals/metalloids (mg/L)

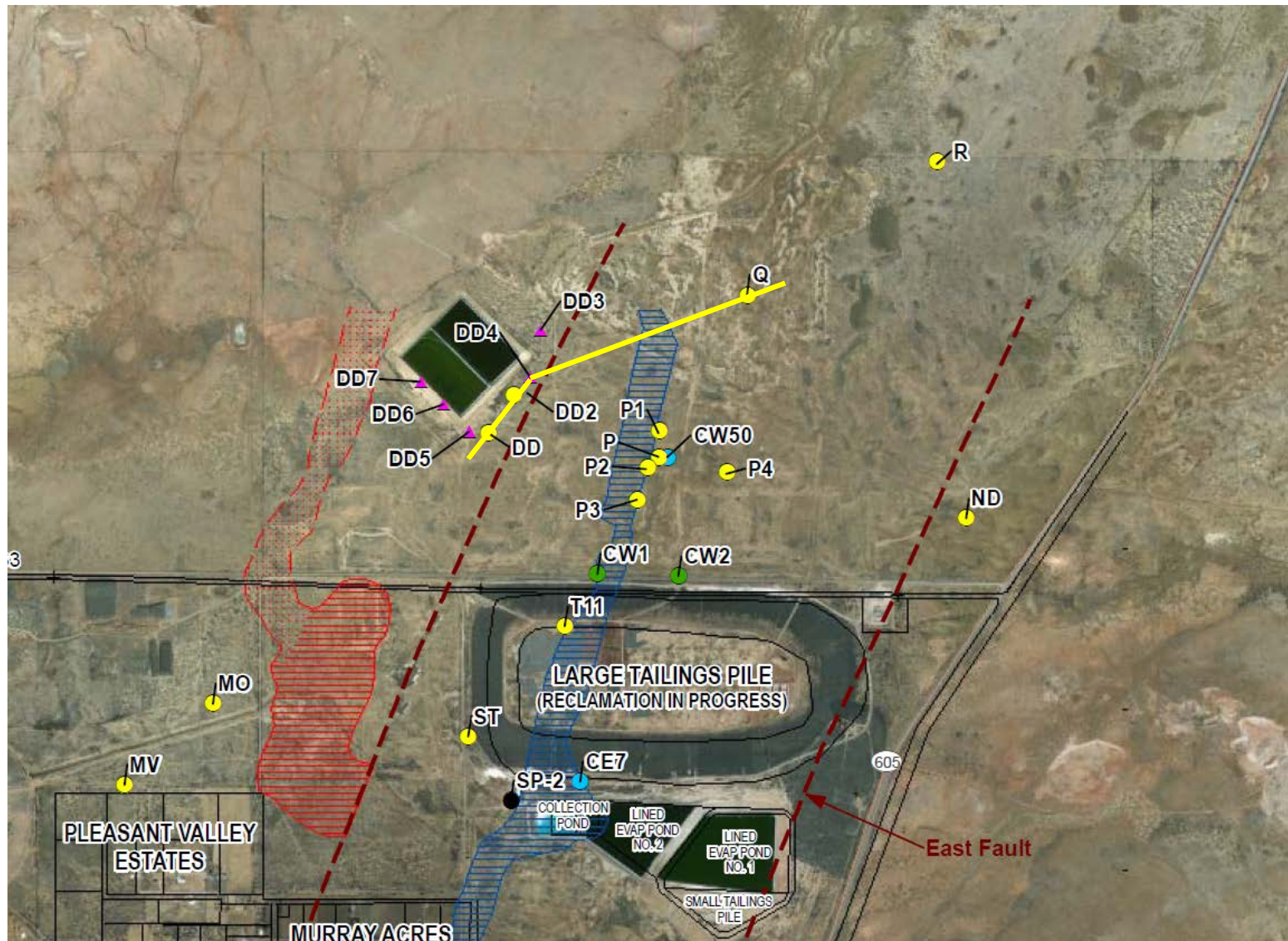


DD2-BK



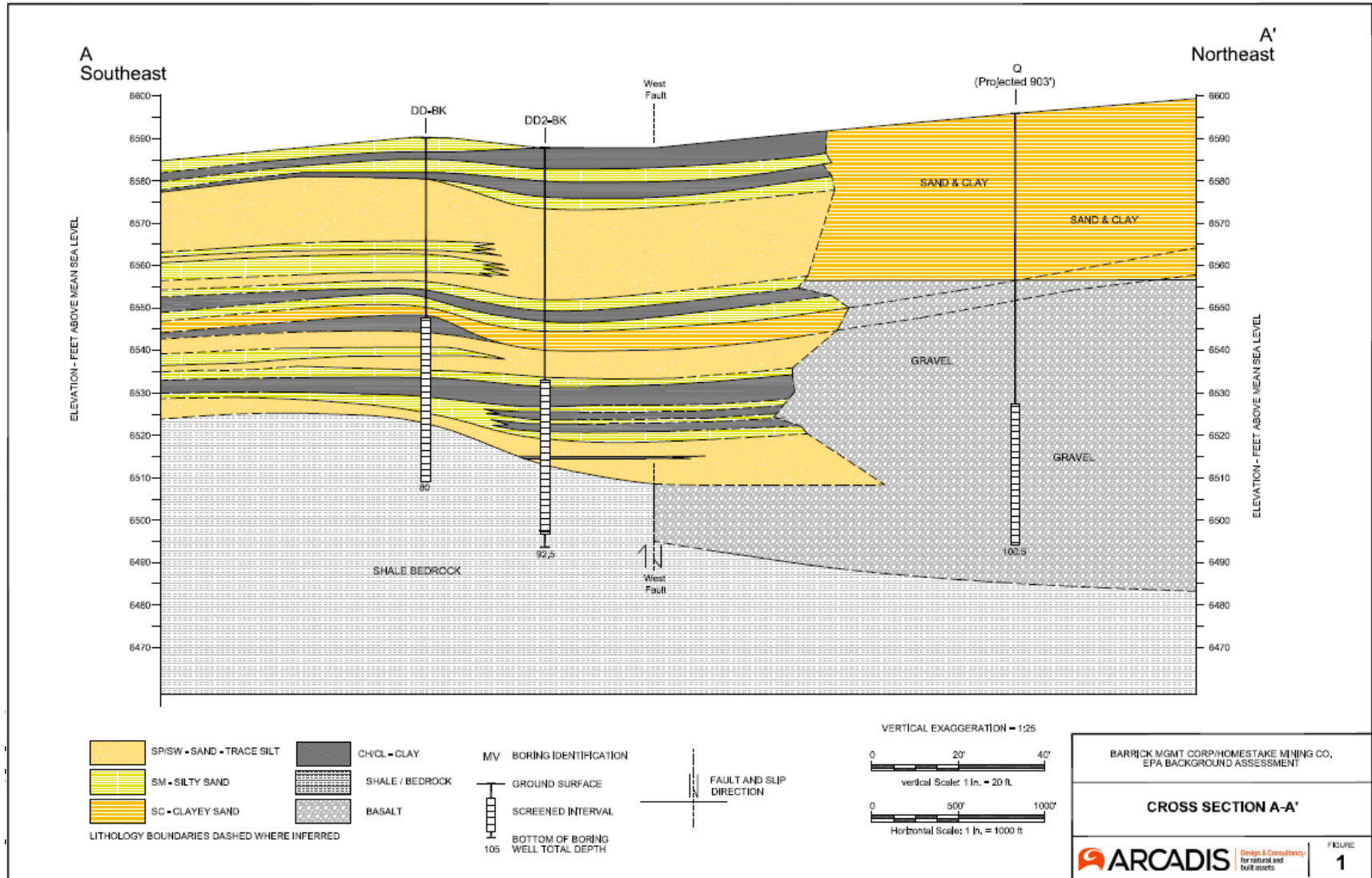
Red circle = non-detect; reporting limit shown
Total Al and Fe/1000; K/100; Na, P/20

Cross section – Location

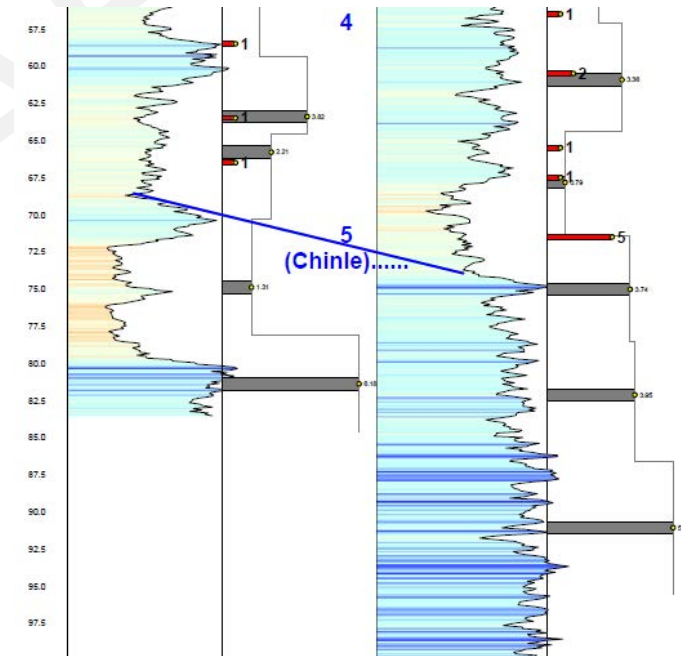
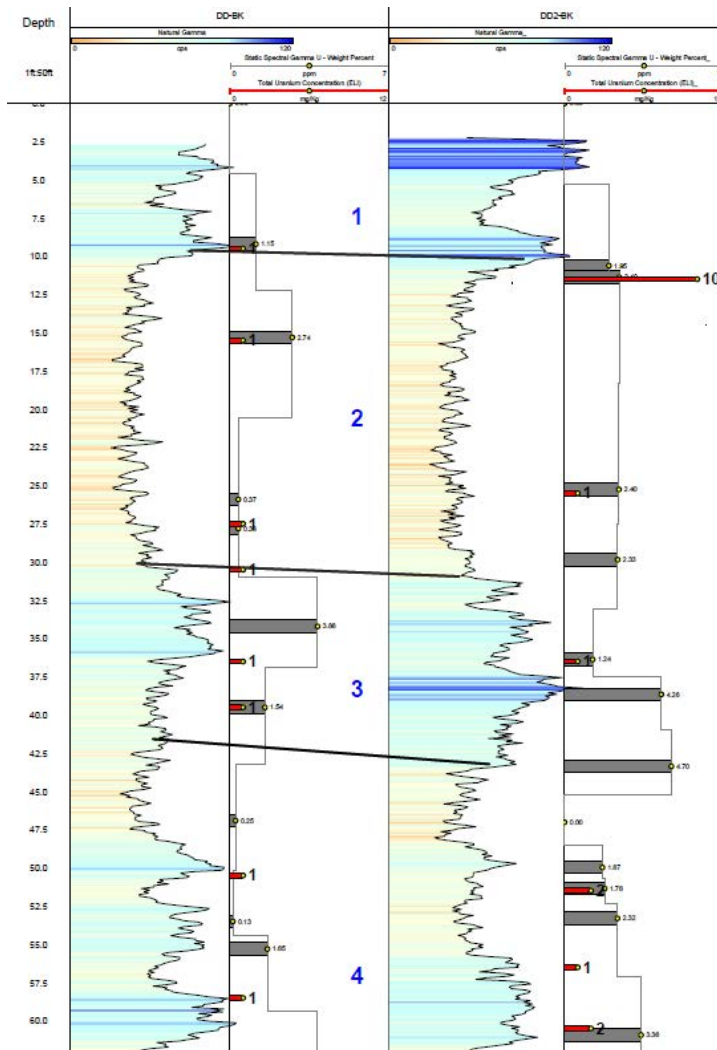


Cross section

New information in this area! Changes the overall interpretation of the DD/DD-2 area and is more consistent with depositional environment as presented by many geologists over last 100+ years

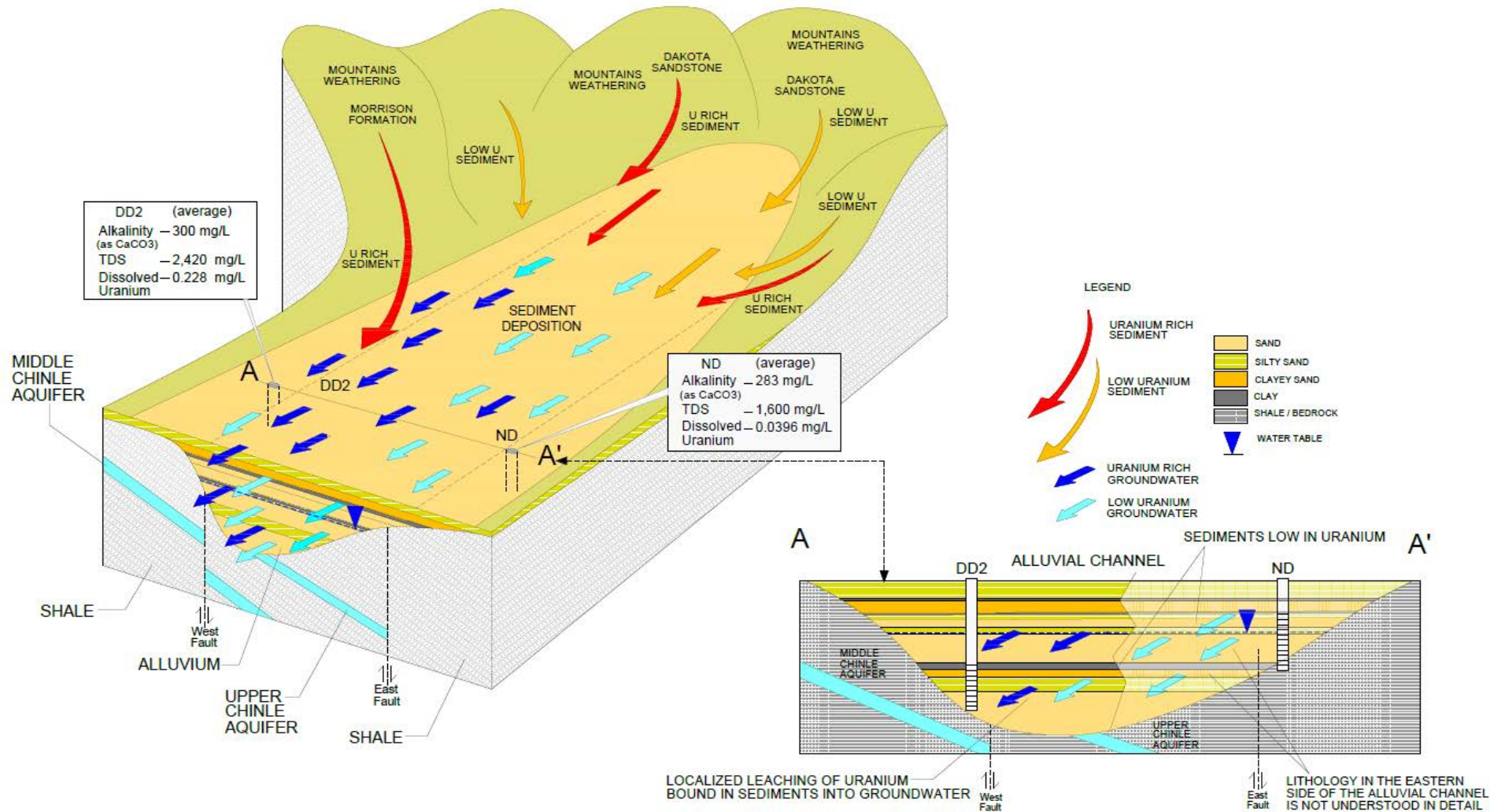


DD-BK and DD2-BK comparison natural gamma logs

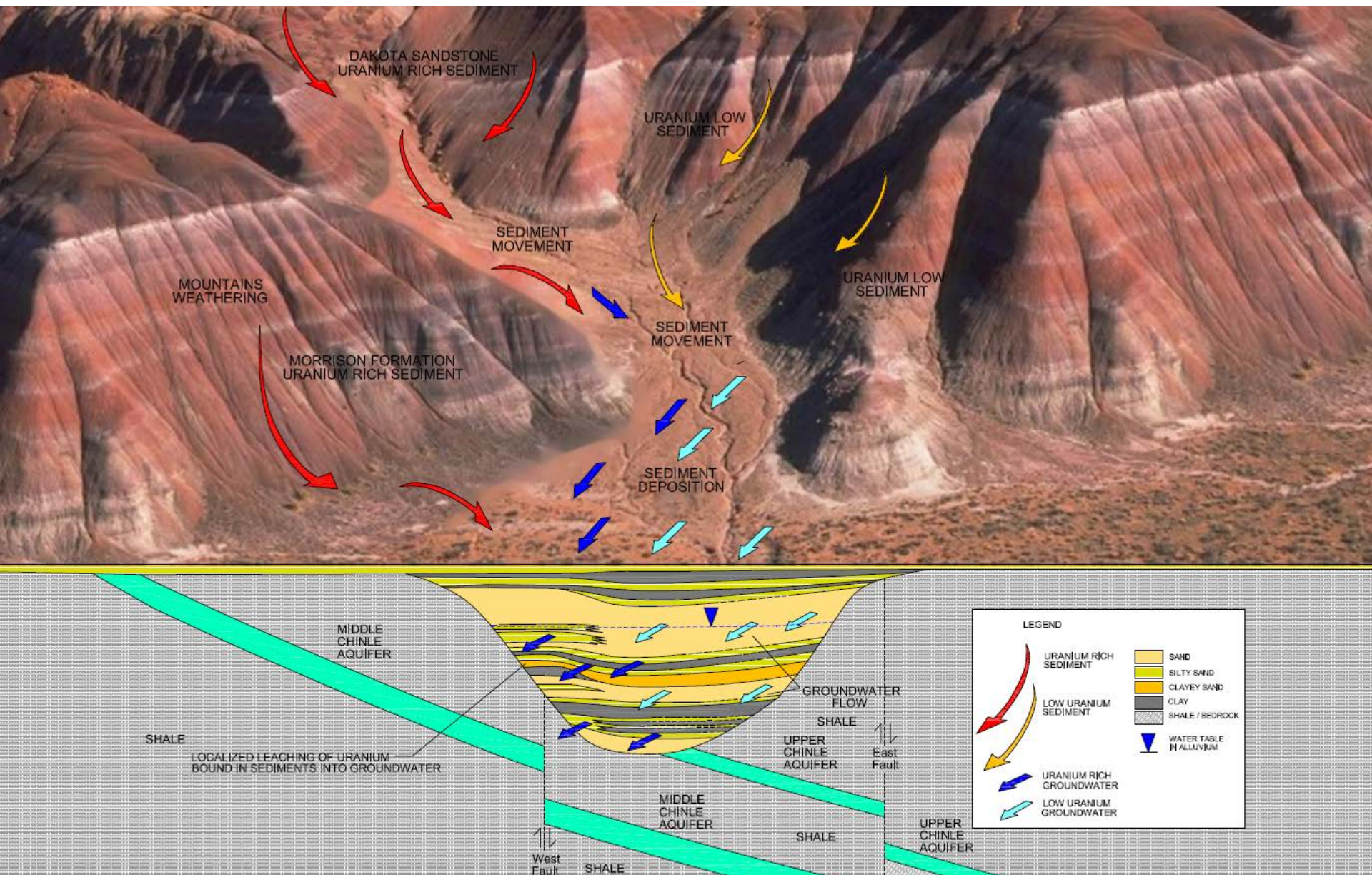


Conceptual Site Model for Naturally Occurring Concentrations of Uranium in Groundwater

CSM and X-Section



CSM Erosion Block



Conclusions

- Passive samplers in 2016 may have not equilibrated
- Geophysical results indicate uranium in alluvium is preferentially in fine-grained sediments and varies by location
- Natural background uranium from alluvial system supported by DD-BK and DD2-BK results
 - Geology is more heterogeneous than previous logs suggested
 - Variation in U concentration in soils between lithologies and also both saturated and unsaturated zones
- CSM developed for uranium migration, placement and groundwater leaching to derive variation in uranium concentrations naturally in basin
- ND well is a geochemical outlier and only represents one point in the alluvial system and as such background uranium assessment should not be based solely on this one well